

Synthesis of a Poly(ϵ -caprolactone) Monosubstituted First Generation Dendrimer by Lipase Catalysis

Armando Córdova,[†] Anders Hult,[‡] Karl Hult,[§] Henrik Ihre,[‡] Tommy Iversen,^{*,†} and Eva Malmström[‡]

Swedish Pulp and Paper Research Institute, STFI
P.O. Box 5604, SE-114 86 Stockholm, Sweden
Department of Biotechnology, Royal Institute of Technology
SE-100 44 Stockholm, Sweden
Department of Polymer Technology
Royal Institute of Technology
SE-100 44 Stockholm, Sweden

Received June 29, 1998

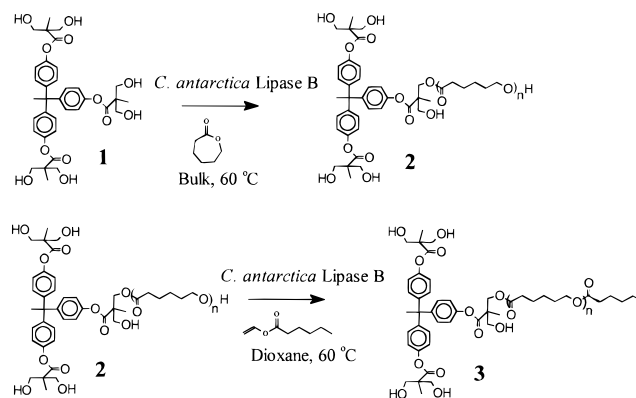
We report the synthesis of a poly(ϵ -caprolactone) (PCL) monosubstituted first generation dendrimer and the selective acylation of the hydroxyl end-group of the PCL chain by lipase catalysis.

Dendrimers are a new type of macromolecules characterized by a well-defined and highly branched layered structure with a multitude of chain ends.¹ Aliphatic polyester dendrimers based on 2,2-bis(hydroxymethyl)propanoic acid (bis-MPA)² are, for example, used as scaffolds in the synthesis of ferroelectric dendrimers³ and as multifunctional initiators for stannous-2-ethylhexanoate (Sn(Oct)₂)-catalyzed ring-opening polymerization (ROP) of ϵ -caprolactone (ϵ -CL) to give controlled synthesis of branched macromolecules.⁴

The aliphatic polyester PCL and its copolymers are of great interest for applications in biological and biomedical areas due to their desirable properties of biodegradability, biocompatibility, and permeability.⁵

Lipase-catalyzed ROP of lactones is a special type of transesterification where no leaving group is released as a separate molecule. It has received increased attention over the past few years, and various polyesters can be produced by the choice of lactone and lipase.⁶ *Candida antarctica* lipase B (CALB) is an efficient catalyst in the ROP of lactones⁷ and regioselective in the acylation of carbohydrates.⁸ Using glycosides as initiators for

Scheme 1. ROP and Selective Acylation of the PCL End-Group



the ROP of ϵ -CL, CALB regioselectively acylates the primary hydroxyl group.⁹ CALB is also efficient in catalyzing the end-group functionalization of PCL.¹⁰

Our aim was to obtain a PCL monosubstituted first generation bis-MPA dendrimer using the hexahydroxy-functional dendrimer **1** (Scheme 1) as an initiator for the ROP of ϵ -CL. The high selectivity of lipase catalysis, in combination with lipase-catalyzed polyester synthesis, is an attractive alternative to poorly selective chemical catalysts.

The hexahydroxy-functional dendrimer **1**, with six equivalent hydroxyl groups, was synthesized according to a procedure developed by Hult et al.^{2b,4a} The dendrimer-initiated ROP of ϵ -CL was performed according to Córdova et al.^{9a,10} The ROP mechanism occurs in two steps. In the nonregioselective initial step, the Ser105¹¹ of the lipase will make a nucleophilic attack on ϵ -CL, and formation of an acyl-enzyme intermediate is achieved.^{6a,7,9} The acyl-enzyme intermediate can then be deacylated by any of the six equivalent hydroxyl groups of **1** to form **2** ($n = 0$) (this deacylation could be stereoselective¹²). The second step, propagation step, was regioselective since the selectivity of the enzyme will decide whether the 6-hydroxy group or the bis-MPA hydroxyl groups of **2** ($n = 0$) will deacylate the acyl-enzyme intermediate. CALB was regioselective for the less sterically hindered 6-hydroxy group of **2**, and hence, a selective chain propagation occurred. The ROP in dioxane gave a low conversion of the dendrimer (20%) and cyclic PCL byproducts. Therefore, ROP was conducted in bulk (Scheme 1). Figure 1 shows a MALDI-TOF MS spectrum from a 24 h ROP of ϵ -CL (200 mg, 1.75 mmol), initiated by **1** (30 mg, 0.05 mmol) and catalyzed by CALB (10 mg).¹³ The average M_w of the PCL substituted products was 2573 Da, with a polydispersity of 1.2 as determined by MALDI-TOF MS. 98% of **1** was converted to PCL substituted products¹⁴ and >99% of the ϵ -CL was consumed.¹⁵ There was a small amount of water initiated PCL, <5% of the consumed ϵ -CL, that

(8) (a) Adelhorst, K.; Björklund, F.; Godfredsen, S. E.; Kirk, O. *Synthesis* **1990**, 112. (b) Córdova, A.; Iversen, T.; Hult, K. *Biotechnol. Lett.* **1997**, *19*, 15. (c) Lay, L.; Panza, L.; Riva, S.; Khitri, M.; Tirendi, S. *Carbohydr. Res.* **1996**, *291*, 197. (d) Woudenberg-van Oosterom, M.; van Rantwijk, F.; Sheldon, R. A. *Biotechnol. Bioeng.* **1996**, *49*, 328.

(9) (a) Córdova, A.; Iversen, T.; Hult, K. *Macromolecules* **1998**, *31*, 1040. (b) Bisht, K. S.; Deng, F.; Gross, R. A.; Kaplan, D., L.; Swift, G. *J. Am. Chem. Soc.* **1998**, *120*, 1363.

(10) Córdova, A. Doctoral Dissertation, Royal Institute of Technology, Stockholm, 1998.

(11) Uppenberg, J.; Hansen, M. T.; Patkar, S.; Jones, A. *Structure* **1994**, *2*, 293.

(12) Wang, Y.-F.; Wong, C.-H. *J. Org. Chem.* **1988**, *53*, 3127.

(13) *Candida antarctica* lipase B, Novozym 435 (7000 PLU/g, PLU = propyl laureate units), an immobilized enzyme, from Novo Nordisk A/S.

(14) Determined by MALDI-TOF MS and TLC, as described in refs 7a and 10.

(15) Determined by GC as described in ref 9a.

* To whom correspondence should be addressed.

[†] Swedish Pulp and Paper Research Institute.

[‡] Department of Polymer Technology.

[§] Department of Biotechnology.

(1) (a) Fréchet, J. M. J. *Science* **1993**, *263*, 1710. (b) Tomalia, D. A.; Naylor, A. M.; Goodard, W. A. *Angew. Chem., Int. Ed. Engl.* **1990**, *29*, 138. (c) Fréchet, J. M. J.; Hawker, C. J.; Wooley, K. L. *J. M. S. Pure Appl. Chem.* **1994**, *A31*, 1627.

(2) (a) Ihre, H.; Hult, A.; Söderlind, E. *J. Am. Chem. Soc.* **1996**, *118*, 6388. (b) Ihre, H.; Hult, A.; Fréchet, J. M. J.; Gitsov, I. *Macromolecules* **1998**, *31*, 4061.

(3) Busson, P.; Ihre, H.; Hult, A. *J. Am. Chem. Soc.* **1998**, *35*, 9070.

(4) (a) Trollås, M.; Hedrick, J. L.; Mecerreys, D.; Dubois, P.; Jérôme, R.; Ihre, H.; Hult, A. *Macromolecules* **1997**, *30*, 8508. (b) Trollås, M.; Hedrick, J. L.; Mecerreys, D.; Dubois, P.; Jérôme, R.; Ihre, H.; Hult, A. *Macromolecules* **1998**, *31*, 2756. (c) Trollås, M.; Hedrick, J. L. *J. Am. Chem. Soc.* **1998**, *120*, 4644.

(5) (a) Vion, J. M.; Jérôme, R.; Teyssy, P. *Macromolecules* **1986**, *19*, 1828. (b) Chiellini, E.; Solaro, R. *Adv. Mater.* **1996**, *8*, 305. (c) Fujisato, T.; Ikada, Y. *Macromol. Symp.* **1996**, *103*, 73. (d) Slomkowski, S. *Macromol. Symp.* **1996**, *103*, 213.

(6) (a) Uyama, H.; Kobayashi, S. *Chem. Lett.* **1994**, 1149. (b) Uyama, H.; Takeya, K.; Kobayashi, S. *Proc. Jpn. Acad.* **1993**, *69B*, 203. (c) MacDonald, R. T.; Pulapura, S. K.; Svirkin, Y. Y.; Gross, R. A.; Kaplan, D. L.; Akkara, J.; Swift, G.; Wolk, S. *Macromolecules* **1995**, *28*, 73. (d) Knani, D.; Gutman, A. L.; Kohn, D. *J. Polym. Sci., Part A: Polym. Chem.* **1993**, *31*, 1221. (e) Uyama, H.; Takeya, K.; Kobayashi, S. *Bull. Chem. Soc. Jpn.* **1995**, *68*, 56. (f) Svirkin, Y. Y.; Xu, J.; Gross, R. A.; Kaplan, D. L.; Swift, G.; Wolk, S. *Macromolecules* **1996**, *29*, 4591. (g) Nobes, G. A. R.; Kazlauskas, R. J.; Marchessault, R. H. *Macromolecules* **1996**, *29*, 4829.

(7) (a) Córdova, A.; Iversen, T.; Hult, K.; Martinelle, M. *Polymer* **1998**, *39*, 6519. (b) Bisht, K. S.; Henderson, L. A.; Gross, R. A.; Kaplan, D. L.; Swift, G. *Macromolecules* **1997**, *30*, 2705.

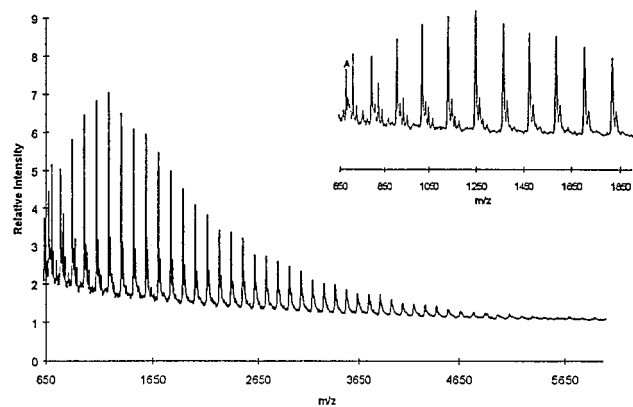


Figure 1. MALDI-TOF MS spectrum of the products from a 24 h bulk polymerization with a DHB matrix. The inset shows an expanded view of the 650–1860 Da region of the spectrum. A = remaining initiator **1**.

was removed by flash chromatography.⁹ MALDI-TOF MS registered detectable peaks from 768 to 6126 Da (corresponding to 1-mer and 48-mer, including the dendrimer end group), with a repeat unit of 114 Da. In this case, MALDI-TOF MS could not determine the degree of substitution of **1**, since the masses will be the same for a PCL hexa- to a PCL monosubstituted dendrimer with the same number of monomer units. Also, ¹H NMR analysis of the reaction mixture did not confirm the degree of substitution or whether it was a mixture of PCL mono-, di- or higher substituted dendrimers. However, ¹H NMR confirmed that **1** was partially substituted, since the aryl protons of the core molecule did not show two symmetrical doublets (Figure S1 in the Supporting Information).^{2a,4a,c}

¹³C NMR was used to investigate whether both of the hydroxyl groups of the same bis-MPA unit of **1** had been acylated. The quaternary bis-MPA carbons of **1** in CDCl₃ are known to shift if they are di- (46.6 ppm), mono- (48.9 ppm) or nonsubstituted (49.5 ppm).^{4a,4c,16} Only two peaks appeared in the spectrum at 48.9 and 49.8 ppm corresponding to PCL monosubstituted and PCL nonsubstituted bis-MPA units, respectively (Figure S2 in the Supporting Information.). This confirms the fact that only PCL monosubstitution of a bis MPA unit of **1** occurred. The selectivity of CALB was used once more to perform a selective acylation of the PCL hydroxyl end-group without further acylation of the bis-MPA hydroxyl groups of **2** (Scheme 1). Figure 3, spectrum 2, shows the result from a 24 h selective acylation of the reaction products from a bulk polymerization (spectrum 1) dissolved in dioxane. The acyl donor was vinyl hexanoate and peaks corresponding to mono- and dihexanoyl acylated products were detected (spectrum 2). The peak area ratio of mono- to diacylated products ($\sum a_{i \text{ mono}} / \sum a_{i \text{ di}}$) was 9:1 and 90% **3** (Scheme 1) was obtained.¹⁷ The average M_w of **3** was 2345 Da, with a polydispersity of 1.3. MALDI-TOF MS registered detectable peaks from 888 to 6111 Da (corresponding to the 1-mer and 47-mer, including the dendrimer and hexanoic acid end groups), with a repeat mass of 114 Da. The diacylated products could either have been di-PCL substituted **1** or mono-PCL substituted **1**, where one of its five bis-MPA hydroxyl groups had been further acylated. MALDI-TOF MS confirmed that mono-PCL substitution had occurred, since the mass was different for **3** compared to the dihexanoyl acylated products. Figure 2, shows the ¹H NMR spectrum of a

(16) Malmström, E.; Johansson, M.; Hult, A. *Macromolecules* **1995**, *28*, 1698.

(17) Estimated by using the peak area ratio of mono- to diacylated products ($\sum a_{i \text{ mono}} / \sum a_{i \text{ di}}$) and extrapolate to calibrated weight ratios, as described in refs 7a and 10.

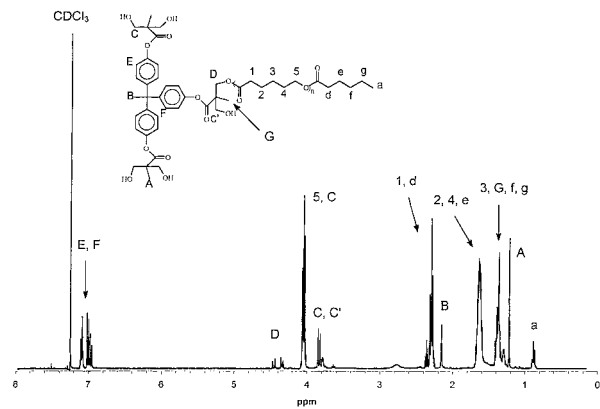


Figure 2. ¹H NMR (400 MHz) of a pure fraction of **3** (Scheme 1) in CDCl₃.

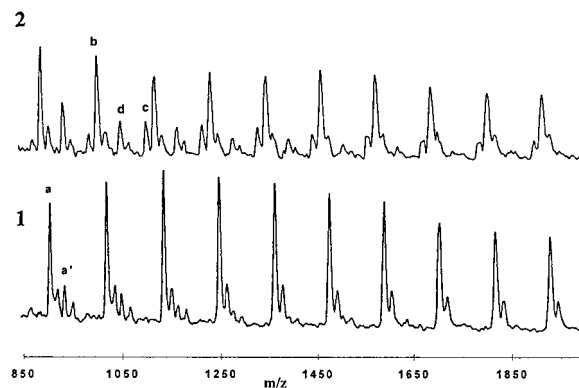


Figure 3. Spectra of the selective acylation of the products obtained in the bulk polymerization, catalyzed by *C. antarctica* lipase B. Spectrum 1 = 0 h. Spectrum 2 = 24 h. a = Na⁺-cationized PCL substituted **1**. a' = Na⁺-cationized linear PCL, b = Na⁺-cationized mono-hexanoyl acylated **2** (Scheme 1), c = Na⁺-cationized di-hexanoyl acylated **2**, d = Na⁺-cationized hexanoate terminated linear PCL.

pure fraction of **3**. It can be seen that the PCL hydroxyl end group was acylated, since there is no corresponding triplet at 3.65 ppm with the same intensity as protons D.^{4c,6c,7a,10} The area ratios of (E + F):D:B:A:a were 12:2:3:6:3 (12ArH, CH₂, CH₃, 2CH₃, CH₃) confirming the mono-PCL substitution of **1**.

In conclusion, selective monosubstitution of hydroxy-functional dendrimers, in combination with polyester synthesis, could be achieved by lipase catalysis. These products would provide numerous opportunities for further modification. The PCL hydroxyl end-group of **2** was selectively acylated by using *C. antarctica* lipase B as the catalyst and gave **3** in an overall yield of 85% from **1**. This synthetic strategy allows selective functionalization of **2**, and it can be used as a macromonomer to prepare various heteromultiarm block copolymer structures. Further studies in this regard are currently underway in our laboratory.

Acknowledgment. Novozym 435 was generously given by Novo Nordisk A/S, Denmark. Financial support from the Swedish National Board for Industrial and Technical Development is gratefully acknowledged.

Supporting Information Available: Experimental Section, Figures S1–S3 and C–H correlation spectrum of **3** (7 pages, print/PDF). See any current masthead page for ordering information and Web access instructions.

JA982252U